

Emission coefficient for line emission  
— Einstein coefficient (approx)

eq 9.8:

$$j_\nu = \frac{n_j A_{ji} h\nu}{4\pi} \phi(\nu)$$

# density  
of atoms w/e- in  
level j

e.g. Gaussian  
eg eq 9.1

most  
commonly

— challenge b/c lots of uncertainties  
(see list section 9.4)

can simplify w/o compromising result too much

3 cases

- radio recombination lines (RRLs)
- optical recombination lines (Balmer)
- $\lambda 21\text{cm}$  neutral hydrogen line

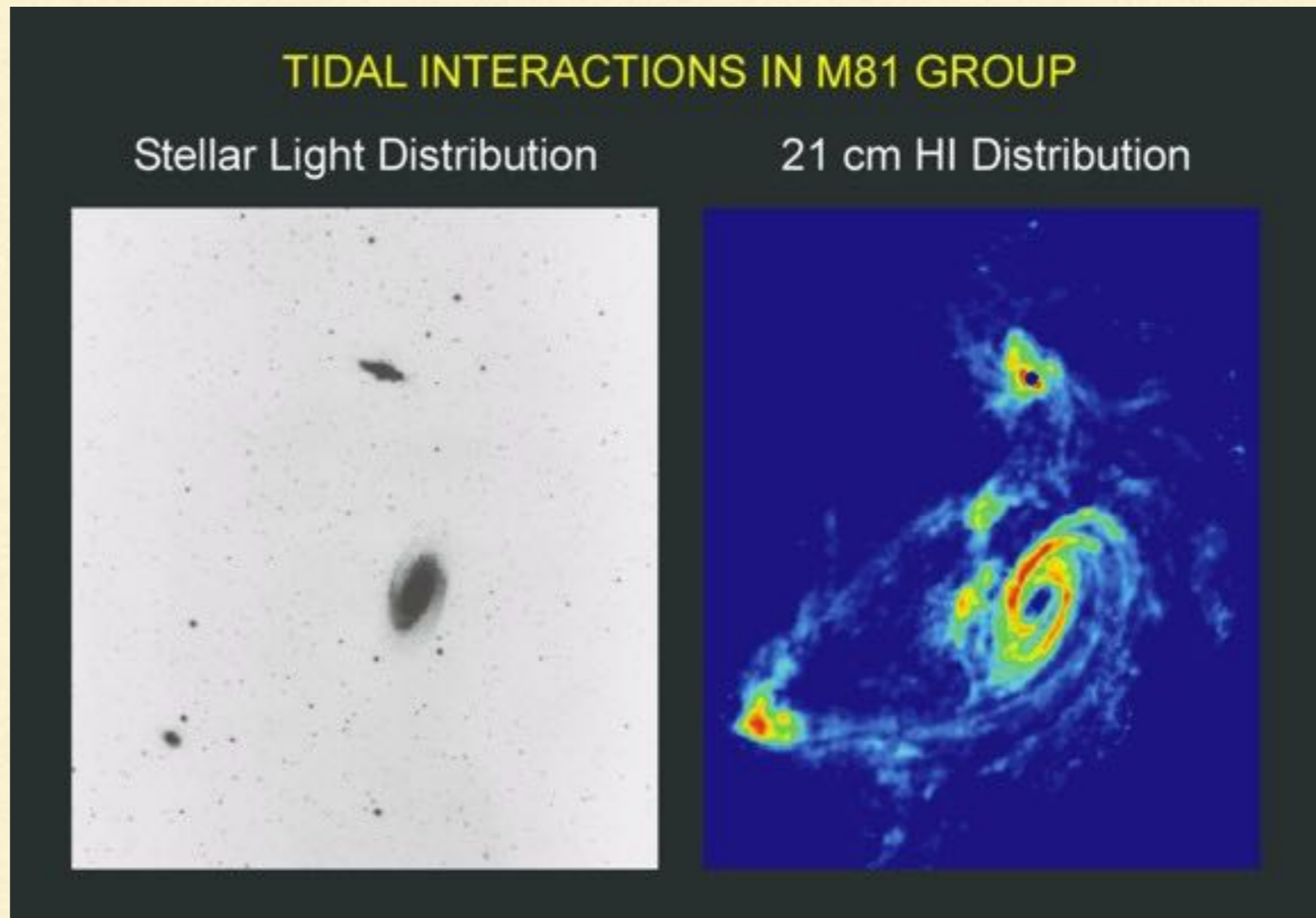
- spin flip transition
- collisionally excited  $\Rightarrow$  LTE, Boltzmann eq 3.23
- bypasses extinction
- everywhere & leads of it

Learned leads

eg. (see 307)

- tidal interactions
- rotation  $\Rightarrow$  DM
- large scale structures

Here, use it to measure HI mass



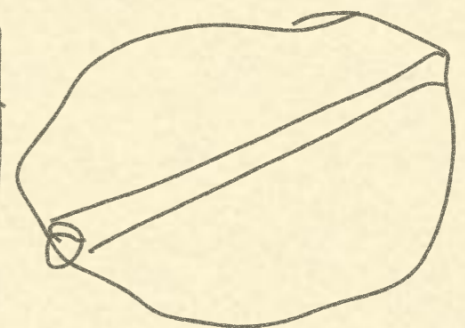


Need introduce column density

definition  $N_{HI} \equiv \int_l n_{HI} dl$  [cm<sup>-2</sup>] eq 9.22

Number density integrated over line-of-sight

can use in  $\tau_\nu = \sigma_\nu \int_l n_i dl = \left[ \frac{n_i}{n_{HI}} = \frac{1}{4} \text{ ; eq 9.20} \right]$

$= \frac{1}{4} \sigma_\nu \int_l n_{HI} dl = \frac{1}{4} \sigma_\nu N_{HI}$  (eq 9.21)  ( $\sigma_\nu =$  effective cross-section)

$\Rightarrow \dots \Rightarrow N_{HI} = 3.85 \cdot 10^{14} T_S \int \tau_\nu d\nu$  (eq 9.23) ( $\tau_\nu \ll 1$ , eq 9.25)

$T_S =$  spin temperature, can assume = kinetic temp  $T$  (when line is in LTE)





Ex 9.4 ugc 2903

$\tau_v \ll 1, \Omega_s = 1.8 \cdot 10^{-5} \text{ sr}$

$\int I_v dv = 0.0458 \text{ Jy/beam km/s}$

beam:  $\Omega_b = 5.21 \cdot 10^{-9} \text{ sr}$

$M_{\text{tot}} \approx 1.6 \cdot 10^{11} M_{\odot}$

$\int I_v dv = 8.79 \cdot 10^{-17} \text{ erg/s/cm}^2/\text{Mz/sr km/s}$

use R-3 eq 4.6  $\Rightarrow \int_v T_B dv = 142 \text{ K km/s}$

$B_v(T) = \frac{2 \nu k T}{c^2}$

$\Rightarrow [9.28] \Rightarrow N_{\text{HI}} = 2.58 \cdot 10^{20} / \text{cm}^2$

mult. by  $N$  with source area  $\Rightarrow$  total # of HI atoms

eq B.2:  $\sigma_s = r^2 \Omega_s [r = \text{distance} = 86 \text{ Mpc}] = \sigma = 1.27 \cdot 10^{46} \text{ cm}^2$  ★

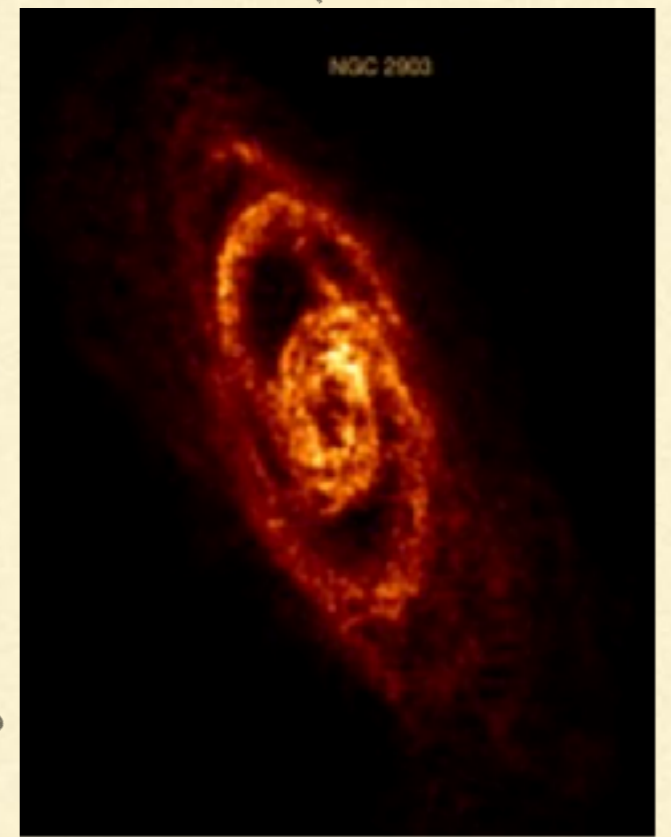
★  $N = N_{\text{HI}} \cdot \sigma_s = 3.27 \cdot 10^{66} =$

total # of HI atoms

x mass of 1 HI atom, divide by  $M_{\odot}$

$\Rightarrow M_{\text{HI}} = 2.7 \cdot 10^9 M_{\odot}$

fraction HI mass 2% of total HI from THINGS



# Optical recombination lines

$$(9.4.2) \alpha_{ij} \ll 1$$

Balmer lines observed in optical

Only need know # of particles in  $n_j$  to use eq 9.8  
(expression for  $j_r$ )

$$\Rightarrow \dots \Rightarrow \int j_r dv = \frac{1}{4\pi} h \nu_{ji} \alpha_{ji} n_e n_i \approx \frac{1}{4\pi} h \nu_{ji} \alpha_{ji} n_e^2$$

use 6.17 with 9.13

effective recombination coeff see table 9.4  
freq of line centre  
e-ion density

$\propto$  density

$$\Rightarrow \dots \Rightarrow \int I_r dv = \frac{1}{4\pi} h \nu_{ji} \alpha_{ji} n_e^2 l = 2.66 \cdot 10^{17} h \nu_{ji} \alpha_{ji} EM$$

set to equal

intensity over the line

$\propto$  temp  $\left[ \frac{pc}{au^6} \right]$



**Table 9.4.** Effective recombination coefficients<sup>a</sup>

$\alpha_{j,i}$ ( $\text{cm}^3 \text{s}^{-1}$ )	Temperature (K)		
	5 000	10 000	20 000
$\alpha_{3,2}$ (H $\alpha$ )	$2.21 \times 10^{-13}$	$1.17 \times 10^{-13}$	$5.97 \times 10^{-14}$
$\alpha_{4,2}$ (H $\beta$ )	$5.41 \times 10^{-14}$	$3.03 \times 10^{-14}$	$1.61 \times 10^{-14}$

<sup>a</sup>Case B recombination has been assumed (see Sect. 9.4.2). H $\alpha$  values are from Ref. [160] and H $\beta$  values are from Ref. [115]. For the latter reference, means for the densities,  $10^2$  and  $10^4 \text{ cm}^{-3}$  are given.

↑ well known, for any 2 lines in an atom,

$E_M$  is the same

$$\frac{\int_{\nu} T_{\nu} d\nu (j \rightarrow i)}{\int_{\nu} T_{\nu} d\nu (k \rightarrow m)} = \frac{\nu_{ji} \alpha_{ji}}{\nu_{km} \alpha_{km}} \propto \text{Temp}$$

Ex H $\alpha$  & H $\beta$

Assume HII region  
T =  $10^4$  K

$$\frac{\nu_{32} \alpha_{32}}{\nu_{42} \alpha_{42}} = \left[ \begin{array}{l} \text{table C.1} \\ \text{table 9.4} \end{array} \right] = \frac{486.132}{656.288} \cdot \frac{1.17 \cdot 10^{-13}}{3.03 \cdot 10^{-14}} = 2.86$$